

**Facile synthesis of (*R*)N-2-hydroxyacyl-L-cysteine derivatives:  
(*R*)N-2-hydroxyacyl transfer from enzymatically-synthesized  
(*R*)S-2-hydroxyacylglutathione derivatives to L-cysteine**

*Short Communication*

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**Summary.** N-(*R*)-2-Hydroxyacyl-L-cysteine derivatives were conveniently synthesized by the reaction of the corresponding S-(*R*)-2-hydroxyacylglutathione with cysteine. The (*R*)2-hydroxyacyl group was transferred from the S-glutathionyl moiety to S-cysteinyl, forming the corresponding (*R*)S-2-hydroxyacylcysteine; this rearranged to the (*R*)N-hydroxyacylcysteine. These compounds have anti-proliferative activity associated with the inhibition of *de novo* pyrimidine synthesis.

**Keywords:** Amino acids – N-2-Hydroxyacylcysteine – S-2-Hydroxyacylglutathione – Glyoxalase – Glutathione – Cysteine

**Abbreviations:** TRIS, tris(hydroxymethyl) aminomethane; DTNB, 5,5'-dithiobis(2-nitrobenzoic acid).

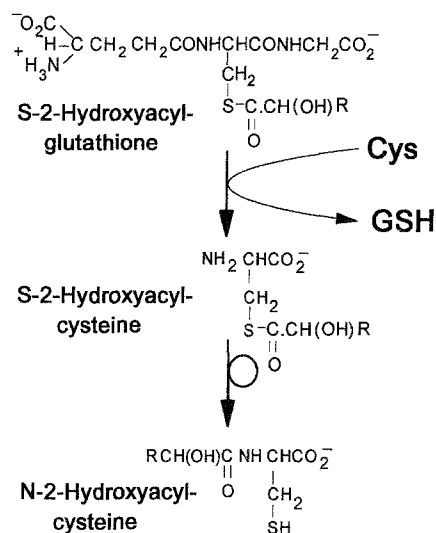
The synthesis of (*R*)N-2-hydroxyacyl-L-cysteine derivatives is normally a complex process involving preparation of the appropriate 2-oxoacyl chloride (Ottenheim and de Man, 1975), reaction of the 2-oxoacyl chloride with L-cystine, reduction of N,N'-bis(2-oxoacyl)-L-cystine to N-(2-hydroxyacyl)-cysteine and separation of the *R*- and *S*- isomers. (*R*)N-2-Hydroxyacyl-L-cysteine derivatives are of interest as intermediates in the extracellular metabolism of S-2-hydroxyacylglutathione derivatives. N-D-Lactoylcysteine is formed from the metabolism of S-D-lactoylglutathione by  $\gamma$ -glutamyl transferase and dipeptidase, and has been implicated in the anti-leukaemia activity of S-D-lactoylglutathione (Edwards and Thornalley, 1994; Clelland et al., 1992). The anti-leukaemia activity is mediated by inhibition of *de novo* pyrimidine synthesis where the prospective mechanism of action is inhibition of the zinc metalloenzyme dihydro-orotase by N-D-lactoylcysteine (Edwards and Thornalley, 1994). Dihydro-orotase activity (EC 3.5.2.3) is associated

with part of the multi-enzyme protein CAD which catalyzes the first three steps of committed *de novo* pyrimidine synthesis (Evans et al., 1993). Dihydro-orotase catalyzes the formation of dihydro-orotate from N-carbamoylaspartate. It is susceptible to inhibition by N-acyl-L-cysteine derivatives which are presumed to proffer a potent thiol ligand to the prosthetic zinc ion and inhibit the enzyme (Christopherson et al., 1989).

A possible approach to the facile synthesis of N-2-hydroxyacylcysteine derivatives was by acyl transfer from the corresponding S-2-hydroxyacylglutathiones. S-2-Hydroxyacylglutathione derivatives were conveniently prepared by enzymatic synthesis from  $\alpha$ -oxoaldehydes and reduced glutathione, catalyzed by glyoxalase I (EC 4.4.1.5) (Clelland and Thornalley, 1991). Previous studies (Tate, 1975) had shown that the *in situ* generation of N-unblocked S-acylcysteinyl derivatives was followed by rapid intramolecular acyl transfer to form the corresponding N-acylcysteinyl derivative. Also, the slow kinetics of the spontaneous hydrolysis of S-2-hydroxyacylglutathione derivatives at pH 7.4 (Reeves and Thornalley, 1993) suggested that it should be possible to achieve reaction conditions where the transfer of the acyl group from S-2-hydroxyacylglutathione to cysteine was favoured over hydrolysis of the thiolester.

(R)N-2-Hydroxyacylcysteine derivatives were conveniently synthesized in high yield by (R)-2-hydroxyacyl transfer from the corresponding (R)S-2-hydroxyacylglutathione to cysteine with rearrangement of the S-(R)-2-hydroxyacylcysteine formed *in situ* to the (R)N-2-hydroxyacylcysteine (Scheme 1). Although the proposed intermediate (R)S-2-hydroxyacylcysteine was not detected, its formation *in situ* is probable given the greater nucleophilicity of the thiol group than that of the  $\alpha$ -amino group of cysteine, and the precedent of the formation of N-D-lactoylcysteinyl-glycine from S-D-lactoylglutathione when the  $\gamma$ -glutamyl residue blocking the cysteinyl  $\alpha$ -amino group was removed by  $\gamma$ -glutamyl transferase (Tate, 1975). Indeed, an alternative procedure for the synthesis of (R)S-2-hydroxyacylcysteine derivatives was a 2-step enzymatic procedure with  $\gamma$ -glutamyl transferase-catalyzed conversion of (R)S-2-hydroxyacylglutathione to (R)N-2-hydroxyacylcysteinyl-glycine, followed by dipeptidase-catalyzed conversion of (R)N-2-hydroxyacyl cysteinylglycine to (R)S-2-hydroxyacylcysteine (Tate, 1975; Edwards and Thornalley, 1994). Having tried the first step with commercial  $\gamma$ -glutamyl transferase however, we obtained only poor yields (<10%) of the (R)N-2-hydroxyacylcysteinylglycine derivative; the majority of the product was reduced glutathione and the corresponding (R)-aldonic acid. This may indicate that commercial  $\gamma$ -glutamyl transferase has thiol esterase activity. Hence, the acyl transfer reaction to cysteine was preferred.

(R)N-2-Hydroxyacyl-L-cysteine derivatives are of interest as prospective inhibitors of dihydro-orotase and they have anti-proliferative activity associated with the inhibition of *de novo* pyrimidine synthesis (Edwards and Thornalley, 1994).



**Scheme 1.** Synthesis of (*R*)N-2-Hydroxyacyl-L-cysteine derivatives

**Table 1.** NMR Spectra of (*R*)N-2-hydroxyacylcysteine derivatives in D<sub>2</sub>O

Compound	N-D-Lactoyl-cysteine	N-D-Mandelyl-cysteine	N-D-Glyceroyl-cysteine
<b>Proton NMR spectra</b>			
Assignment $\delta$ ( <i>J</i> )			
Cysteinyl			
2-H	4.45	4.48	4.50
3A-H	2.88	2.88	2.88
3B-H	2.81	2.81	2.82
( <i>J</i> <sub>2,3A</sub> )	2.9	4.7	5.1
( <i>J</i> <sub>2,3B</sub> )	6.6	6.5	6.4
( <i>J</i> <sub>3A,3B</sub> )	−14.3	−14.0	−14.2
Other			
	Lactoyl	Mandelyl	Glyceroyl
	2-H = 4.15	2-H = 5.07	2-H 4.19
	3-H(3H) = 1.21	Ph = 7.26	3-H(2H) 3.68
	( <i>J</i> <sub>2,3</sub> = 7.0)		( <i>J</i> <sub>2,3</sub> = 4.1)
<b><sup>13</sup>C NMR spectra</b>			
Assignment $\delta$			
Cysteinyl			
C-1	175.6	172.9	173.2
C-2	56.5	54.0	54.3
C-3	27.3	29.6	25.1
Other			
	Lactoyl	Mandelyl	Glyceroyl
	C-1 180.0	C 1 174.4	C-1 174.5
	C-2 69.9	C-2 73.2	C-2 72.3
	C-3 21.9	C-1(Ph) 137.8	C-3 63.3
		C-2,6(Ph) 128.4	
		C-3,5(Ph) 126.6	
		C-4(Ph) 126.5	

Proton chemical shifts  $\delta_{\text{H}}$  (ppm) and coupling constants *J*<sub>Hx,Hy</sub> (Hz), and carbon-13 chemical shifts  $\delta_{\text{C}}$  (ppm) of spectra recorded at 270 MHz and 68 MHz, respectively, of 10 mM (*R*)N-2-hydroxyacylcysteine derivatives (10 mM), pD 1.6.

### Materials and methods

N-D-Lactoylcysteine was synthesised by the reaction of S-D-lactoylglutathione (200 mg, 100 mM) with L-cysteine (150 mM) in TRIS/HCl buffer (50 mM), pH 7.4 and 25 °C. The reaction progress was followed by withdrawal of small aliquots of reaction mixture (5 µl) and assay for thiol groups with 5,5'-dithiobis(2-nitrobenzoic acid) (D TNB); as the reaction proceeds, the thiol concentration was expected to increase from 150 to 250 mM. The reaction reached completion after 10 min. Similar reactions were performed with S-L-glyceroylglutathione and S-D-mandelylglutathione and gave N-L-glyceroylcysteine and N-D-mandelylcysteine, respectively. The product mixture was lyophilized to dryness and reconstituted with 10 mM HCl (4 ml). The N-2-hydroxyacylcysteine derivative was separated from residual reactants and reduced glutathione co-product by loading of the product mixture on a column (2.6 cm × 30 cm) of Dowex 50 cation exchange resin, proton form, and elution with 10 mM HCl, 72 ml/h. The eluate was monitored at 254 nm. Residual S-2-hydroxyacylglutathione and cysteine, reduced glutathione and TRIS buffer were retained on the column, and the N-2-hydroxyacylcysteine derivative was eluted in the retention volume range 144–240 ml. The product fractions were lyophilized to dryness to give the N-2-hydroxyacylcysteine derivatives in yields: 77% (N-D-lactoylcysteine), 68% (N-L-glyceroylcysteine) and 63% (N-D-mandelylcysteine). The products were characterized by <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy (Table 1), FAB mass spectrometry and TLC analysis. The FAB mass spectra of the (*R*)N-2-hydroxyacylcysteine derivatives gave M+1 peaks of m/z 194 (N-D-lactoylcysteine), 210 (N-L-glyceroylcysteine), and 256 (N-D-mandelylcysteine). TLC analysis was on silica gel with developing solvent propan-1-ol:acetic acid:water, in the volume ratio 10:5:5, and detection by ninhydrin in ethanol (0.2%). Chromatographic R<sub>f</sub> values were: 0.33 (N-D-lactoylcysteine), 0.52 (N-L-glyceroylcysteine), and 0.58 (N-D-mandelylcysteine).

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### References

- Christopherson RI, Schmalzl KJ, Szabados E, Goodridge RJ, Harsanyi MC, Sant ME, Alagar EM, Anderson JE, Sharma SC, Bubb WA, Lyons SD (1989) Mercaptan and dicarboxylate inhibitors of hamster dihydroorotase. *Biochemistry* 28: 463–470
- Clelland JD, Thornalley PJ (1991) S-2-Hydroxyacylglutathione derivatives. Enzymatic preparation, purification and characterisation. *J Chem Soc Perkin Trans I*: 3009–3015
- Clelland JD, Allen RE, Thornalley PJ (1992) Inhibition growth of human leukaemia 60 cells in culture by S-2-hydroxyacylglutathiones and their monoethyl ester derivatives. *Biochem Pharmacol* 44: 1953–1959
- Edwards LG, Thornalley PJ (1994) Prevention of S-D-lactoylglutathione-induced inhibition of human leukaemia cell growth by uridine. *Leuk Res* 18: 717–722
- Evans DR, Bein K, Guy HI, Lui X, Molina JA, Zimmerman BH (1993) CAD gene sequence and the domain structure of the mammalian multifunctional protein CAD. *Biochem Soc Trans* 21: 186–191
- Ottenheijm HC, De Man JHM (1975) Syntheses of α-keto acid chlorides. *Synthesis*: 163–174
- Reeves ME, Thornalley PJ (1993) The hydrolysis of S-D-lactoylglutathione. *Biochem Soc Trans* 21: 169S

Tate SS (1975) Interaction of  $\gamma$ -glutamyltranspeptidase with S-acylglutathione derivatives. FEBS Lett 54: 319–321

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